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Utilization of copolymer based on poly (vinyl alcohol) and 2-ethylhexyl acrylate as admixture for cement pastes

Abstract: In this study, three composition ratios of [polyvinyl alcohol (PVA)- 2-ethylhexyl acrylate (2-EHA)] were prepared by grafting polymerization techniques in aqueous solution using ammonium persulfate as free radical initiator. The ratio of PVA/2-EHA in the grafted copolymers was variable: 1/ 8, 1/ 6 and 1/ 4, respectively. The obtained grafted copolymers were characterized by different techniques including FTIR, ¹H NMR, TGA and DSC. The copolymers were mixed with cement to study the physicomechanical properties of cement pastes including W/C ratios, setting time, workability, water absorption as well as compressive strength were investigated. The results showed that the addition of grafted copolymers to cement pastes affected the physicomechanical properties. As the content of 2-EHA in the grafted copolymers increased, the water of consistency decreased, whereas the setting times (initial & final) were shortened. The compressive strength of the cement pastes was increased at all ages of hydration while water absorption decreased with increasing 2-EHA.

Key words: PVA, 2-EHA, copolymer, admixture, cement, workability.

Introduction

Admixtures are ingredients in concrete that mixed with water, cement, and aggregates to produce concrete with high workability, durability, and compressive strength. Admixture can be classified by their function, including air-entraining, water-reducing, plasticizers, accelerating, retarding, hydrationcontrol and miscellaneous admixtures [1-3]

PVA is one of the major polymers used in the industry because of its excellent chemical resistance and physical properties [4]. PVA has been widely used in a broad range of applications in different industries such as for textile sizing, adhesives, protective colloids for emulsion polymerization and plasticizer for cement [5]. Styrene/methacrylate copolymer has been prepared by emulsion polymerization in presence of PVA as surfactant. The obtained polymer was mixed with cement to study physical and mechanical properties of cement pastes and mortars [6, 7]. The results showed that the emulsion copolymers in presence of PVA obviously improved most of the specific characteristics of the cement pastes [6]. The addition of PVA to cement pastes, significantly improved flexural strength, permeable pore volume and water absorption as well as water/ cement ratio [8]. Negim et al., [9] have incorporated a modified PVA by blending with carbamide as a chemical admixture to cement pastes.

Singh and Sarita [10] reported that PVA increased the strength and decreased the porosity due to the interaction of PVA with cement. On the other hand, Negim et al have prepared acrylate copolymers including poly(2-hydroxy ethyl acrylate-co-styrene) [11], poly(2-hydroxy ethyl acrylate-co-butyl acrylate) [12], poly(methyl cellulose-co-poly(acrylic acid)[13], poly(acrylic acid-co-styrene)[14], poly(2-hydroxyethyl acrylate-co-vinyl ether of ethylene glycol) [15], poly(methoxypolyoxyethylene-g-polymaleic anhydride) [16], poly(acrylic acid-co-butyl methacrylate)[17], poly(2-hydroxy ethyl acrylate-co-2-hydroxyethyl methacrylate)[18] and poly (methyl methacrylate-co-ethylhexyl acrylate-co-methacrylic acid) [19]. The effect of acrylic copolymers on prop-

erties of cement pastes was investigated. The addition of acrylic copolymers to cement pastes improved the properties of cement pastes such as compressive strength, water absorption, setting time as well as workability.

This article describes the preparation of acrylic copolymer based on polyvinyl alcohol (PVA) and 2-ethylhexyl acrylate (2-EHA) by grafting polymerization. The effect of variable ratios of PVA/2-EHA on the physico-mechanical properties of cement pastes was studied.

Materials and methods

Materials

Polyvinyl alcohol (PVA), 2-ethylhexyl acrylate (2-EHA) and ammonium persulfate (APS) were obtained from Sigma Aldrich, USA. All these chemicals were of pure grade and used as received.

The raw materials used in the present study are Portland cement clinker (PCC) and raw gypsum (G). Each of those raw materials was separately ground in a steel ball mill until the surface area of respectively 3650 and 2800 cm²/g was achieved. The chemical composition of the raw materials is shown in Table 1. The mineralogical composition of the PCC sample is C_3S , 58.79 %; β - C_2S , 17.68 %; C_3A , 8.08 %; C_4AF , 9.72 %. The Portland Cement (PC) was prepared by mixing 96 % PCC and 4 % G (by weight) in a porcelain ball mill for one hour using 3 balls to ensure complete homogeneity of the cement. The Blaine surface area of the cement sample was 3350 cm²/g [20].

Table 1 – The Chemical Composition of the Raw Materials (M0), Mass %

Oxides Materials	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	MgO	SO ₃	Na ₂ O	K ₂ O	L.O.I
PCC	21.48	6.03	4.22	64.29	0.68	0.39	0.21	0.11	1.32
G	0.58	0.14	0.11	30.08	0.13	45.36	0.07	0.09	22.16

Synthesis of the P(PVA-g-2-EHA)

Graft polymerization was carried out in a 250 mL three- necked flask equipped with a thermometer, reflux condenser, and stirrer. PVA was dissolved in distilled water and stirred continuously at a constant temperature of 60 °C. After the PVA was fully dissolved, the temperature of the system was strictly maintained at a required value. Freshly prepared ammonium persulphate APS solution was added followed by dropwise addition of 2-EHA. The reaction was conducted for two hours with stirring continued for another 20 min at room temperature. The grafting polymerization was prepared with various ratios of PVA /2-EHA (1/8, M1; 1/6, M2 and 1/4, M3). The obtained copolymer were reprecipitated several times from ethanol to diethyl ether and dried in vacuum desiccators at 30 °C until a constant weight was achieved and poly 2-ethylhexyl acrylate dissolved.

The graft yield (G%), the grafting efficiency (GE%) and the amount of homopolymer (H%) formed were calculated according to the following equations:

Graft yield $(G\%)=[(W_1 - W_0)/W_0] \times 100$ Homopolymer $(H\%)=[(W_2 - W_1)/W_3] \times 100$ Grafting efficiency $(GE\%)=(W_1/W_2) \times 100$ Where W_0 , W_1 are the weights of initial matrix and grafted matrix i.e. weight of the product after extraction, respectively. W_2 is the crude product before extraction and W_3 is the weight of monomer [21]. The grafting efficiency of the copolymer was 91.5% for M1, 89.9% for M2 and 83.8% for M3.

Characterization

Chemical composition of the prepared copolymer was verified by FTIR and ¹H NMR spectra, which was recorded on a Bruker Tensor 37 FTIR spectrometer and ¹H NMR (Bruker Avance 300 and 400, Switzerland).

The prepared grafted copolymers were added to the mixing water and then added gradually to 300 g of the dry cement in order to determine the water of consistency and setting time of the cement using Vicat apparatus [21, 22]. The determined water of consistency premixed with the copolymers (M1, M2 and M3) was added to 500 g of the dry cement. The resulting cement pastes were directly moulded into one-inch cube stainless steel moulds. The moulds were manually agitated for 2 minutes and then on a vibrator for another 2 minutes to assure the complete removal of air bubbles and voids and to produce suitable pastes. The moulds were kept in a humidity chamber at 100 % R. H and a constant room temperature overnight, then demoulded and cured under water till the time of testing (1, 3, 7 and 28 days) for total porosity and compressive strength [23]. The compressive strength test was carried out using a hydraulic testing machine of Type LPM 600 M1 SE-IDNER (Germany) having full capacity of 600 KN. The loading was applied perpendicular to the direction of the upper surface of the cubes. The total porosity, ξ of each sample at any interval was calculated from the following equation [24]: where 0.99 is the specific volume of the free water, We is the evaporable water content, dp is the bulk density, g/cm³ and Wt is the total water content which is equal to the sum of evaporable water (We) and combined water (Wn) contents. The bulk density (dp) was determined from the following equation: dp = W1 / W1-W2 g/cm³, where W1 is the saturated surface dry weight in air (g) and W2 is the submerged weight in water (g).

Results and discussion

 $\xi = 0.99 \text{ X We X dp} / (1 + \text{Wt})$

The grafting polymerization in presence of APS including initiation, propagation and termination steps is shown in Scheme 1.



3. Termination



Scheme 1 - Mechanism of grafting polymerization of 2-EHA onto PVA

FTIR spectra

Evidence of grafting formation is provided by FTIR spectroscopy. From the FTIR spectrum of PVA (Fig. 1a), it is obvious that PVA shows a broad peak at 3300 cm⁻¹ for the –OH group, and one peak at 1087 cm⁻¹ attributed to –C-O- groups as in the study reported in ref. [25]. FTIR spectrum of the grafted copolymer is shown in Fig. 1b. It can be observed C–H broad alkyl stretching band (n = 2850-3000 cm⁻¹). 1460 cm⁻¹ and 1380 cm⁻¹ suggest the bending vibration of C–H. The C-O stretching at approximately 1159 cm⁻¹, is attributed to the ether (C-O). The peak at 1732 cm⁻¹ represents the C=O stretching band and the peak at 1150 cm-1 is attributed to C-O-C are distinguished.

¹H NMR spectrum

The ¹H NMR spectrum of grafted copolymer dissolved in chloroform is shown in Fig. 2. The assignment peaks of the proton species in PVA and its grafted with 2-EHA. The typical peaks for grafted copolymer are indicated in Fig. 2 appeared at δ = 0.87-0.89, 1.27-1.4, 1.66, 1.88, 2.3, 3.32, 3.74 and 4.48 ppm correspond to the protons bonded to C1, C2, C3, C4, C5, C6, C7 and C8 respectively [25-31].

Thermal properties

The thermal behaviour of pure PVA and P(PVA-g-2-EHA) was evaluated with TGA, DTG in air at a heating rate of 10°C/min and DSC at the same heating rate under nitrogen atmosphere. The TGA thermograms for the pure PVA and P(PVA-g-2-EHA) are given in Fig. 3. TGA measurements of pure PVA and grafted PVA with 2-EHA in different compositions indicate that thermal degradation temperature is higher than 400 °C as shown in Table 2. TGA curves also indicate that, the pure PVA follows five step degradation process as shown in Fig. 3 and Table 2. The first weight losses occur between 30 and 200 °C, which corresponds to the removal of water. The second, third, fourth and fifth weight loss occurs between 200 to 900 °C correspond to completely decomposition with loss of CO₂. Fig. 3 and Table 2 show the lost weight % of grafted copolymer M1. One stage decomposition was observed between 29 and 910 °C, while decomposition of grafted copolymer M2 was observed in 2 steps. The first step ranging between 29 and 317°C, and the second step of degradation occurs between 317 and 907 °C. The rate of degradation in the second step is slightly faster than that in the first step.



Figure 1 – FTIR spectra of (a) PVA and (b) P(PVA-g-2-EHA)





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Three weight losses were observed in grafted copolymer M3 as shown in Fig. 3 and Table 2. The first weight loss was observed in the range 29 - 176 °C in step 1 of degradation, which corresponds to the loss of water. Upon heating PVA above the decomposition temperature the polymer begins a rapid chainstripping elimination of water [26, 27]. The second weight loss was observed in between 176 and 333 °C in step 2 of degradation. The third weight lost was between 333 and 905 °C and it was much shorter than the first. The maximum polymer degradation temperature (PDT_{max}) (28, 29) corresponds to the temperature at which the maximum rate of weight loss occurred and appeared 460 °C for pure PVA blend and at 425 ° C for pure grafted copolymers M1, M2 and M3 [30, 31].

Glass transition temperatures (T_g) of pure PVA and grafted polymer with different ratios of 2-EHA were calculated from the corresponding typical DSC traces as given in Fig. 4 and Table 2. The T_g value for pure PVA is 70 °C as shown in Table 2. From the data present in the DSC curves, it can be seen that, the increase of 2-EHA content from 20 to 40% shifts the glass transition to higher temperatures. M1 has higher T_g value due to the higher 2-EHA content. No data was observed from the second heating cycle for all the formulation.



Figure 3 – TGA thermograms of PVA and P(PVA-g-2-EHA) (M1-M3)

Table 2 - Thermal properties of PVA and PVA-g-2-EHA

Polymer Ratio %	$T_g(^{\circ}\mathrm{C})^{\mathrm{a}}$	Temperature (°C) for decomposition ^b	Weight loss ^b (%)	Residual ^b (%)	PDT(°C) ^b
PVA	70	30-200 200-433 433-454 454-511 511-900	5.9 68.1 4.875 12.15 8.81	94.01 25.9 21.031 8.87 0.006	460
M1	60.74	29.34-910	100	0.0	425
M2	46.9	29-317 317-907	19.5 80	80.5 0.0	425

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Continuation of table 2

M3	45.2	29-176 176-333 333-905	7.2 24.6 68	92.7 68.1 0.0	425
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^a Determined from DSC curves

^b Determined from the derivative of TGA.

Application of Graft Copolymer for Cement

Water/ Cement Ratio

The results W/C ratios of pure cement (M0) and various cement pastes mixed with grafted copolymers (M1-M3) are shown in Fig. 4. It is clear that the W/C ratio of cement pastes decreases from 0.312 to 0.29 with addition of the grafted copolymers. However, W/C ratios increased with decreasing 2-EHA content in the grafted copolymers. These W/C ratios of cement maybe change according to kind of polymers, and concentrations of polymer during the mixing of cement with polymers (1, 31).

Setting time

The setting time (initial & final) of cement pastes premixed with grafted copolymers are given in Fig. 6. The setting time of cement pastes is extended when mixed with grafted copolymers. However, the initial and final setting time of cement pastes shorten with increasing the content of 2-EHA in the grafted copolymers. For example, initial and final setting time of M1 is lower than those of M2 and M3. This attributed to the evaporation of water from the surface of the cement pastes mixed with (M1) is less than those of cement pastes mixed with M1 and M2 [32, 33].



Figure 4 – DSC thermograms of PVA and P(PVA-g-2-EHA) (M1-M3)

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Figure 5 – Effect of P(PVA-g-2-EHA) content on the W/C ratios of cement pastes



Figure 6 – Effect of P(PVA-g-2-EHA) content on the setting time of cement pastes.

Workability

The results of the cement pastes workability are presented graphically in Fig. 7. The results showed that cement paste had an initial workability of 135 mm; however, there was an increase in workability measurement with the addition of grafted copolymers. This increase in workability was attributed to surface active of the grafted copolymers [34]. Fig. 7 shows an increase in flow of mortar with increasing the content of 2-EHA in the grafted copolymers.



Figure 7 – Effect of P(PVA-g-2-EHA) content on the flow of cement paste

Water absorption

Generally, the contact of cement with water, cement paste absorbs water because of its porous microstructure. The lower the water absorption, the lower the permeable pore volume [34].

The effect of grafted copolymers with different content of 2-EHA on water absorption of hardened cement paste is shown graphically in Fig. 8. It can be seen that water absorption decreased with addition grafted copolymers to cement pastes. This indicates that the addition of polymer results in reduction of the porosity of the pastes [33, 34]. However, the water absorption of cement pastes mixed with grafted polymers decreased with increasing of 2-EHA in the grafted copolymers.

Compressive strength

The effects of grafted copolymers on the compressive strength of cement pastes are represented as a function of curing time in Fig. 9. As shown from Fig. 9, the compressive strength of cement pastes increased with addition of grafted copolymers to cement pastes at nearly all-curing periods. Furthermore, an increasing in the 2-EHA content in the grafted copolymers increased the compressive strength of the cement pastes. Cement pastes mixed with 40% 2-EHA (M1) gave highest strength with 78.7 MPa while cement pastes mixed with 20% 2-EHA (M3) gave lowest strength with 71.9 MPa. This is attributed to the agglutinant action of grafted copolymers with grains of cement and led to improve the mechanical properties of cement pastes (35).



Figure 8 – Effect of P(PVA-g-2-EHA) content on water absorption of cement pastes.



Figure 9 – The effect of P(PVA-g-2-EHA) on the compressive strength of cement pastes

Conclusions

Grafted copolymers of polyvinyl alcohol (PVA) with 2-Ethylhexyl acrylate (2-EHA) in three different ratios, of 1:8, 1:6 and 1:4 were prepared and characterized in relation to FT-IR, ¹H NMR, TGA and DSC. From thermal analysis, it is shown that, the degradation profiles of grafted copolymers obtained from aqueous dispersion were influenced by the ratios of 2-EHA in the grafted copolymers. The rate of degradation for the grafted polymers is thermally stable formulation depending upon the 2-EHA loading from 30 to 900 °C. Transition glass temperature (T_g) was determined from DSC. The value of T_g shifted from 45 °C to 60 °C with increasing the content of 2-EHA in grafted polymers and indicated that the grafted copolymers are miscible.

Mixing of cement pastes with water and the grafted polymer P(PVA-g-2-EHA) evidently improves the properties of cement pastes. The W/ C-ratio decreases, i.e., the grafted polymer acts as a water reducing agent when mixed with the cement pastes. Setting time of cement pastes increased with decreasing the content of 2-EHA. The workability of cement pastes also enhances and seems to be higher than that of the pure Portland cement pastes. Water absorption of the hardened cement pastes premixed with the grafted copolymers decreased, while compressive strength increased compared with those of the reference cement pastes. As the ratio of 2-EHA in the grafted polymer increased, the properties of the cement pastes also improved.

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